



St. Lawrence High School

A JESUIT CHRISTIAN MINORITY INSTITUTION
First Term Examination -2019



Subject – Physical Sc.

Class - VII

Full Marks : 90

SECTION - A

Fill in the Blanks :

1X5=5

1. A **Molecule** of an element is made up of atoms of same kind.
2. Photosynthesis is **Chemical** Change
3. A molecule of a **hydrogen** gas element contains one atom.
4. Volume of a cubical box of side 14 cm is **$14 \times 14 \times 14 = 2744$** cm³
5. Motion of smoke particles is an example of **Random** motion.

Choose the correct option

1. Weight of a body of mass 1kg is
a. **1kgf** b. 10kgf c. 100kgf d. none of these
2. The speed of very fast moving objects is measured in
a. m/s b. cm/h c. **km/h** d. m/min
3. If the valency of chlorine is 1 what is the valency of copper in CuCl₂ ?
a. 1 b. **2** c. 3 d. 4
4. The greater the intermolecular space, the _____ is the cohesion.
a. Greater b. **Lesser** c. Same d. None of these
5. In which if the following reaction does carbon dioxide react with moisture of water ?
a. Respiration b. **Photosynthesis** c. Dissolution d. Rusting

State whether the followings statements is true or false

1. A compound Cannot be split into simpler substances by chemical means. (**False**)
2. Dissolution of glucose is a chemical change. (**False**)
3. Valency of HCO₃ is 2. (**False**)
4. Area of a circle with radius r is $2\pi r^2$ (**True**)
5. On complete to & fro motion of the pendulum about its extreme position is called 1 oscillation. (**True**)

Name the following

1. What is the process of conversion of a substance from gaseous state to liquid state called ?
Ans. : Condensation
2. What is the name of the process in which a sugar solution changes into alcohol ? Ans.: Fermentation
3. Name a Triatomic gas. Ans. Ozone
4. Name the type of motion that repeats itself at fixed interval of time . Ans. : Periodic
5. Name the internal volume of a container. Ans : Capacity

Match the following :

1. Non - compressible	a. Random motion
2. Motion of a swing	b. Gas
3. Copper	c. Solid
4. Largest thermal expansion	d. Oscillatory motion
5. Motion of a flying mosquito	e. Variable valency

Ans. : (1)-(c) , (2)-(d), (3)-(e), (4)-(b), (5)-(a)

SECTION - B

Very short answer types question.

1. A girl is running at a constant speed of 4m/s, find the time taken to cover a distance of 90m.

Ans. : Given , Speed = 4m/sec.

$$\begin{aligned} & \text{Distance} = 90\text{m} \\ \therefore & \text{ We known speed} = \frac{\text{Distance}}{\text{Time}} \\ = & \text{ Time} = \frac{\text{Distance}}{\text{Time}} \\ = & \frac{90}{4} = 22.5 \text{ sec.} \end{aligned}$$

2. Why does the weight of a body decreases as we go higher ?

Ans. : The weight of a body decreases as we go higher because gravity decreases as we go higher.

3. Define Speed. Write the SI unit of Speed.

Ans. : Speed is defined as the distance travelled by an object for unit time. SI unit of speed is metre per second (m/s).

4. Name two substances which sublime at room temperature .

Ans. : Naphthalene / Camphor / Iodine / Ammonium Chloride.

5. What do you mean by Endothermic process ? Give an example.

Ans. : The process in which energy / heat is used to complete the reaction are called endothermic process. Eg. dissolution of glucose.

Short answer type question (answer any five)

1. A bus starting from Bangalore covers a distance of 275 km in 5 hrs on Friday evening & returns to Bengaluru in 3 hrs early next morning.

$$\text{Ans. : Average Speed} = \frac{275 + 275}{5+3} = 68.3 \text{ km / 8hr.}$$

2. Explain why is motion of a bicycle a complex motion ?

Ans. : The motion of bicycle performs both rotatory and translatory motion. The wheels of a bicycle performs rotatory motion and the bicycle itself performs translatory motion. So it is complex motion.

3. Explain why burning of a candle is both a physical & chemical change ?

Ans. : Burning of a candle shows following changes ____.

1) The wax under the wick gets heated and melts. The molten wax solidifies while flowing down.

This change is physical.

2) A part of molten wax vaporises and burns to form carbon dioxide and water vapour. This is irreversible change and so chemical change.

4. What is Slaking of lime ?

Ans. : Slaking of lime -

When a very small amount of water is added to the quick lime, the water quickly vaporises out with a hissing sound. With a larger amount of water, a thick white paste of slaked lime is formed. This process is called slaking of lime.

5. What is the valency of the underlined element or radical ?

a. CaCO_3 b. Fe_2O_3 c. Cu_2O

Ans. : (a) 1 (b) 3 (c) 2

6. Write down the chemical formula of the following

a. Ammonium sulphate b. Sodium hydroxide

Ans. : (a) $(\text{NH}_4)_2\text{SO}_4$ (b) NaOH

7. Differentiate between Rectilinear & Curvilinear Motion with suitable example.

Ans. :

Rectilinear

Curvilinear

1. Motion along a straight line.

2. Eg. a car moving along a straight road.

1. Motion along a curved line

2. Eg. a car taking turn on a curve on the road.

SECTION - C

IV. Long answer question (any 8)

1. Describe one activity to show the thermal expansion of a solid, a liquid and a gas.

Ans. : Thermal Expansion :-

To Show the thermal expansion of a solid, a liquid and a gas.

1) Procedure : We place two wooden blocks of same height in front of a wall. We place a metal rod in between the wall and block as shown in the diagram (fig1) we place a pencil along the free end of the rod. We heat the rod.

Observation : We will observe that the rod expands and pointer pencil rotates.

2) We fix a transparent straw to a narrow mouthed bottle and fill with coloured water and seal the mouth. Now we place the bottle in a bowl of hot water as in fig (2).

Observation : We will observe that level of the water in the straw rises as water expands on being heated.

3) We fit an uninflated balloon to the mouth of a bottle. We place the bottle in a pan containing water and heat it.

Observation : We will observe that the balloon will get inflated as air in the bottle expands.

Diagram :

2. Explain an example to show that physical and chemical changes can occur together.

Ans. : Burning of a candle shows following changes_____.

1) The wax under the wick gets heated and melts. The molten wax solidifies while flowing down.

This change is physical.

2) A part of molten wax vaporizes and burns to form carbon dioxide and water vapour. This is irreversible change and so chemical change.

3. How would you measure the volume of an irregular solid which is insoluble in water using a measuring cylinder. Explain with diagram. Also write the principle behind this method.

Ans. : To find density of an irregular solid.

Procedure - We first find the mass (M) of the irregular solid say a pebble using a physical balance .

Now to find the volume of the solid we follow the steps .

(i) We take a measuring cylinder and fill it partially with water

(ii) We note the initial volume of water as V_1 .

(iii) We tie the solid with the string and immerse it completely into the water in measuring cylinder.

(iv) We note the new level as V_2 of water.

(v) We find $V_2 - V_1 = V$, as volume of solid.

: Density of irregular solid $D = M/V$

4. Write 5 differences between mass and Weight.

Ans. :

Mass

1. Amount of Matter contained in a body.
2. Constant quantity.
3. Mass can never be zero.
4. S.I. unit is kg.
5. Beam or physical balance measuring device.

Weight

1. Force with which the earth attracts a body towards centre.
2. Varies from place to place.
3. Weight can be zero if no gravity acts.
4. S.I. unit is N
5. Spring balance measuring device.

5. Define and give one example for each.

Ans. : (i) Vibratory motion :- The to and fro or back and forth in a fixed pattern about its mean position while the rest of the object ultimately remains fixed is called vibratory motion.

Eg. : string of a sitar is plucked.

(ii) Multiple Motion - The motion which is a combination of two or more types of motion is called multiple motion.

eg. : A drill machine used by a carpenter / a bicycle.

(iii) Random Motion - The motion in which an object changes its direction frequently is called random motion.

Eg. : Motion of flying kite.

(iv) Circulatory motion - The motion of an object about a fixed point along the circumference of a circle, without changing its position is called circulatory motion.

Eg. : A spinning top.

(v) Non-periodic motion - The motion which does not repeat itself at fixed intervals of time is called non-periodic motion.

Eg. : motion of the wheels of a car moving on a busy road.

6. Describe an activity to determine the time period of a simple pendulum .

Ans. : To determine the time period of a simple pendulum.

Procedure : We tie a silk thread to the metallic bob. We displace the bob from its mean position so that the simple pendulum starts oscillating. Using stop watch we record the 20 oscillation time taken. Let this time be = t sec.

: Time period of simple pendulum. $T = t/20$ second.

7. Explain how atoms, molecules and radicals differ from each other. Give 1 example for each of them.

Ans. :

<u>Property Independent</u>	<u>Atom</u>	<u>Molecule</u>	<u>Radicals</u>
1. Existence	Not Capable	Capable	Capable
2. Charge	Not Charged	Not Charged	Charged
3. Valency	Has a Valency	Has no Valency	has a valency
4. Number of elements	Only one	More than one	more than one
5. Reacting nature	Directly	Breaks into atoms individually react in	Take part as a unit
6. Example	Hydrogen (H)	Reaction Oxygen (O ₂)	Sodium Iron (Na)

8. What is slaking of lime? What type of change occurs during this process? Explain.

Ans. : Slaking of lime -

When a very small amount of water is added to the quick lime, the water quickly vaporizes out with a hissing sound. With a larger amount of water, a thick white paste of slaked lime is formed. This process is called slaking of lime.

9. Explain the liquid gas interconversion on the basis of intermolecular force. What do you mean by liquefaction?

Ans. : When a liquid is heated to a particular temperature, it forms bubbles, called boiling. At this point, a liquid gets converted fast into gaseous state. The molecules move still far apart. The intermolecular force becomes too weak and a gaseous state results. Eg water boils at 100°C. When a gas is cooled it will change to the liquid below its boiling point, called condensation. Here molecules of a gas become less energetic and come closer. This is called Liquefaction.

10. Differentiate between concave meniscus and convex meniscus.

Ans. :

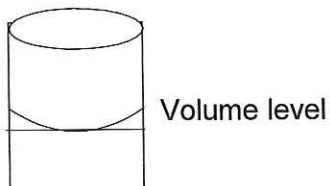
Concave Meniscus

1. The Surface forms curve of liquid downwards when filled in a container is called concave meniscus.

2. Water

3. It wets the sides of container

4.



5. Reading of lower part is taken

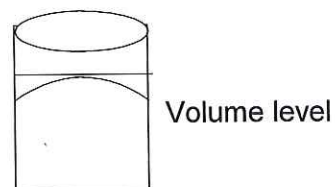
Convex Meniscus

1. The Surface forms curve of liquid upwards is called convex meniscus.

2. Mercury

3. It does not stick to sides of container

4.



5. Reading at the upper most level is taken