



# ST. LAWRENCE HIGH SCHOOL

A JESUIT CHRISTIAN MINORITY INSTITUTION



**Sub: Physical Science**

**Class: 8**

**Date: 25.01.21**

**Duration: 40 min**

**Worksheet Solution 04**

**Full Marks: 15**

## MATTER

### Choose the Correct options:

- 1) Change of state from vapour to liquid is called  
(a) Condensation (b) Liquefaction (c) **Both (a) and (b)** (d) None of these
- 2) The liquefaction point of pure steam is  
(a) 0°C (b) 32°C (c) 4°C (d) **100°C**
- 3) During condensation heat is \_\_\_\_\_ by the substance  
(a) **Lost** (b) Gained (c) Created (d) None of these
- 4) Which of the following is a sublimate?  
(a) Iodine (b) Napthalene (c) Ammonium Chloride (d) **All of these**
- 5) Process through which a vapour directly changes to a solid  
(a) Solidification (b) Deposition (c) **Both (a) and (b)** (d) None of these
- 6) Carbon dioxide changing to dry ice is an example of
- 7) (a) **Solidification** (b) Sublimation (c) Both (a) and (b) (d) None of these
- 8) Melting point of aluminium is  
(a) **660.3°C** (b) 66.3°C (c) 6.63°C (d) 100°C
- 9) The melting point and freezing point of gold is  
(a) 2970°C (b) **1064°C** (c) Both (a) and (b) (d) None of these
- 10) Many germs are killed off by  
(a) **Boiling water** (b) Freezing water (c) Ice (d) Dry Ice
- 11) Boiling point and condensation point of pure water are  
(a) **Same** (b) Different (c) Variable (d) Undetermined
- 12) Another name for liquefaction is \_\_\_\_\_  
(a) fusion (b) Liquefaction (c) **Condensation** (d) Deposition
- 13) The addition of salt to ice cream helps to \_\_\_\_\_ the melting point of ice.  
(a) Raise (b) **Lower** (c) Stabilize (d) Determine
- 14) Heat absorbed during sublimation point is used to increase \_\_\_\_\_  
(a) Kinetic energy (b) Temperature (c) **Intermolecular space** (d) Intermolecular force
- 15) What is the lowest temperature attained with a salt and ice mixture?  
(a) 10°C (b) 21°C (c) **-21°C** (d) 0°C
- 16) The temperature remains constant during the process of \_\_\_\_\_  
(a) Solidification (b) condensation (c) **Both (a) and (b)** (d) Combustion