



Class - 7

ST. LAWRENCE HIGH SCHOOL

A JESUIT CHRISTIAN MINORITY INSTITUTION

Subject- Physical science AnswersofWorksheet- 13

Date -23.04.2020

Chapter- Elements, Compounds and Mixtures (Chemistry)

<u>Q Answer the following questions (MCQ) : (1×15):</u>

Question 1

Q.

Two or more elements or compounds NOT chemically combined. Mixtures

Question 2

Q. A pure substance containing only one kind of atom.

answer choices

- ElementsQuestion 3
 - Q.

The properties of the new substance are usually different than the properties of the elements it contains. answer choices

Compounds

- Question 4
 - Q.

This substance cannot be separated by physical means. Separating this substance requires a chemical reaction. answer choices

Compounds

Question 5

Q.

Over 100 existing of these substances are listed and classified on the periodic table. answer choices Elements

Question 6

Q.

This substance cannot be separated by physical means. Separating this substance requires a chemical reaction. answer choices Compounds

• Question 7 Q. The properties of this substance is similar to the properties of its components. answer choices Elements

Question 8

Q. No reaction between substances. answer choices Mixtures

Question 9
Q.

These substances can be separated into their components by chemical or physical means. answer choices Elements

Question 10

Q. The atoms are chemically combined in some way. answer choices Compounds Question 11

Que: Q.

> These substances cannot be separated into simpler materials (except during nuclear reactions). answer choices Elements

- Question 12
 - Q.

What kind of mixture is uniform and are known as solutions? answer choices homogeneous

 Question 13 Q.
What kind of mixture is non-uniform? answer choices

mixture of compounds

- •
- Q14. NaHCO₃ is an example of —

answer choices

a compound

Question15
Q.

Which are examples of a mixture?

answer choices

Air

Teacher- Piyali Halder