ST. LAWRENCE HIGH SCHOOL



A JESUIT CHRISTIAN MINORITY INSTITUTION

WORKSHEET_52(CLASS-12)

TOPIC- ALDEHYDE AND KETONE



SUBJECT – CHEMISTRY DURATION – 30 mins

F.M. - 15

DATE -17.08.20

1.1 Stephen's red	uction is used to	convert an alkyl	nitrile into-
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- a) Aldehyde b) Ketonec) Alcohol d) Carboxylic acid
- 1.2 Ketones are reduced to-
- a) Primary alcoholb) Secondary alcohol c) Tertiary alcohold) Not possible
- 1.3 HIO4 can oxidise-
- a) Ketoneb) Ether d) Vic di old) gem di ol
- 1.4Rosenmund's reduction is used to prepare aldehyde from-
- a) 4-heptanoneb) Acetonec) Alkyl nitriled) Alkanoyl chloride
- 1.5 Dry distillation of calcium acetate results in the formation of-
- a) Formaldehydeb) Acetaldehydec) Methaned) Acetone
- 1.6In Wacker process, along with PdCl2, air the other reagent used-
- a) Fe2O3 b) None of these c) Fe2O3 d) CuCl2
- 1.7 Acetaldehyde can be identified by-
- a) All of these options are correct b) Tollen's reagent c) Brady's reagent d) lodoform reaction
- 1.8 lodoform test can be used to distinguish between-

a) Ethanol and Ethanalb) Acetaldehyde and Acetone c) Acetone and diethyl ketoned) All of the above 1.9 Which of the following will not give addition reaction with CH3MgBr in Dry ether? a) HCHOb) CH₃CHOc) CH₃CH₂CHOd) CH3OCH3 1.10 On treating acetaldehyde with I2 and NaOHa) Iodoform is obtained b) Black coloured precipitate is formedc) Yellow coloured has is formed d) Methanoic acid is obtained 1.11 Which of the following is a terminal functional group? a) Alkene b) Ether c) Alcohol d) Carboxylic acid 1.12Aldehyde and ketone can be protected and deprotected bya) Reactionwithalcohol b)Reactionwith water c) hydrolysisd) Reaction with alcohol and dry HCl 1.13Pinacolone is aa)Ketoneb) Viccinal di ol c) geminal di ol d) Ester 1.14Chemical reaction of Aldehyde and ketone with H2O leads to formation ofa) alcohol b) stable Viccinal diolc) unstable geminal diol d) Stable product 1.15 Tertiary alcohols on reaction with Cu at 573K producesa) Aldehyde b) ketone c) Alkyne d) Alkene PREPARED BY: MR. ARNAB PAUL CHOWDHURY