

ST. LAWRENCE HIGH SCHOOL

A JESUIT CHRISTIAN MINORITY INSTITUTION

Class: 8 Date: 08.04.20 **Sub:Physical Science Duration: 40 mins** Worksheet 2 **Full Marks: 15** PHYSICAL AND CHEMICAL CHANGES/PHYSICAL CHANGE **Choose the Correct options:** 1. Thermal expansion generally occurs when a substance is (a)Heated Ans (b)Cooled (c)Inflated (d) Pressurized 2. An inflated balloon kept near a hot bulb bursts. Why? (c)Chemical reaction Ans (a)Thermal (b)Melting (d) Insects bite expansion 3. Which theory explains change of state? (a) Kinetic theory (b) Dalton's atomic (c) Bohr's theory (d) Conservation Ans theory theory 4. Which energy of the molecules decreases on cooling? (a) Electrical (b) Kinetic energy (c) Potential energy (d) Light energy Ans energy 5. Which of the following is not a sublimate? Ans (a) Naphthalene (b) camphor (c) Chlorine (d) iodine 6. The substance that dissolves in a solution Ans (a) solute (b) solvent (c) solution (d) suspension 7. Dry ice is used to form artificial smoke because it is (a) a sublimate (b) has high © Has low melting (d)Is abundant Ans melting point point in nature 8. Salt dissolves in water and the volume (a) Increases (b) decreases (c) remains unchanged (d) becomes Ans zero 9. Kinetic energy changes during Ans (a) Expansion (b)Contraction (c) Rise in temperature (d) Change of state 10. A physical change is generally Ans (a) Reversible (b) Irreversible (d) Continuous (c) Random 11. Breaking of a glass is an example of Ans (a) Physical (b) Chemical (c) Reversible change (d) Desirable change change change 12. When solute is dissolved in a solvent it is a (b)Reversible (a)Physical change (c) Irreversible change (d) Both (a) and change (b) 13. Energy is released in (a) Melting Ans (b)Boiling (c) freezing (d) sublimation 14. Tearing a piece of paper is Ans (a) Physical change (b)Chemical (c) Irreversible change (d) Both (a) and change (c) 15. Physical change always (a) Is reversible (b) has no new (c) does not involve (d)Both (a) and Ans substance formed energy change (b)