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ST. LAWRENCE HIGH SCHOOL



(1×15):

A JESUIT CHRISTIAN MINORITY INSTITUTION

Subject- Physical science Answers of Worksheet- 13 Class - 7

Date -24.04.2020

Chapter- Elements, Compounds and Mixtures

Answer the following questions (MCQ):

Q1.All matter consists of particles called_______.
B. Atoms
2. Atoms consist of parts called protons, neutrons, and_______.
A. Electrons
3. Each proton has a _______ electrical charge.
C. Positive
4. Each electron has a _______ electrical charge.

	Negative
•	5
	Each neutron has an electrically charge. C. Neutral
•	6.
	The nucleus of an atom contains protons and
0	C. Neutrons
•	7.
	Electrons move around the nucleus.
0	B. Outside
•	8.
	An element is determined by the number of (also known as its atomic number).
0	B. Electrons
•	9.
	The modern periodic table is organized by atomic number.
0	A. Increasing
	10.
•	IV.

A is a c	combination of two or more atoms that are held		
together by covalent bonds.			
C. Molecules			

- Q11: An element can exist as either atoms or
- A. molecule
 - Q12: The non-metals exist in two of the three states of matter i.e. gas and solid, at $\bf A$. room temperature
- Q13: The elements that does not conduct heat and electricity very well are known as A. non-metal
- Q14: In steel and iron industries, the gas used for cutting, welding and melting of metal is ${\bf A}.$ helium
- **Q15:** The most common type of sugar is A. glucose

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