

Question 4

Q. What is the molecular formula if the empirical formula is CH_2O and the molecular molar mass is 180.18?

answer choices

$\mathsf{CH}_2\mathsf{O}$

 $C_2H_4O_2$

 $C_4H_8O_4$

$C_6H_{12}O_6$

Answer C₆H₁₂O₆

Question 5

Q. What is the empirical formula for the following molecular formula: C_5H_{12}

answer choices

 C_5H_{12}

 CH_3

 CH_2

 $C_{2.5}H_{6}$

Answer C_5H_{12}

Question 6

Q. Which pair has the same empirical formula?

answer choices

NaCrO₄ and Na₂Cr₂O₇

 $C_2H_4O_2$ and $C_6H_{12}O_6$

 $C_3H_6O_3\,and\,C_2H_6O_2$

CH₄ and C₂H₆

Answer $C_2H_4O_2$ and $C_6H_{12}O_6$

Question 7

Q. A 4.50g sample of which of the following would have the greatest mass of oxygen?

answer choices

Na₂O (molar mass = 62g/mol)

Li₂O (molar mass = 30g/mol)

MgO (molar mass = 40g/mol)

SrO (108g/mol)

Answer Li_2O (molar mass = 30g/mol) Question 8

Q. A compound is determined to contain 14g of nitrogen and 32g of oxygen. The empirical formula of the compound is: answer choices NO

 NO_2

 N_2O

NO₃

Answer NO₂

Question 9

Q. How many moles of carbon are in 88g of propane, $C_{3}H_{8}$ answer choices 2.0
16.0
6.0
96.0
Answer 6.0

Question 10

Q. 2H₂O

How many H ₂ O molecules are present?
answer choices

	1	
	4	
	2	
8	8	
An	Answer 2	

Question 11 •

Q. What do we call a chemical substance made up of 2 or more different types atoms bonded together?

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answer choices

formula

atom

subscript

compound

Answer subscript

Question1 2

Q. All ______ are made up of two or more atoms bonded together.

answer choices

atoms

molecules

electrons

elements

Answer molecules

Question 13

Q. Bonds between atoms in a molecule form as a result of the sharing of?

answer choices

formulas

electrons

atoms

chemicals

Answeratoms

Question 14

Q. A recipe for a chemical substance is called a ______

.

answer choices

coefficent

chemical formula

Oxygen

Atom

Answer chemical formula Question1 5

Q. $H_2O_2 \leftarrow$? This number tells the number of atoms present.

answer choices coefficient

molecule

subscript

atom

Answer subscript

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