



St. Lawrence High School  
A Jesuit Christian Minority Institution



Term : 2<sup>nd</sup>

Work Sheet – 34

Subject – Physics

Class – XI

Chapter – Heat

Date – 28.11.20

Choose the correct option for the following questions.

1 × 15 = 15

1. 0.540kg of ice at 0°C is mixed with 0.540kg of water at 80°C. The final temp of the mixture is
  - a. 40°C
  - b. 79.9°C
  - c. 80°C
  - d. 0°C
2. 0.1kg of ice at -10°C is heated till it is converted to 30°C. the amount of heat required is
  - a. 58800J
  - b. 69300J
  - c. 105000J
  - d. 48300J
3. 0.1kg of ice at 0°C is mixed with .1kg water at 100°C. the final temp of the mixture is
  - a. 10°C
  - b. 30°C
  - c. 20°C
  - d. 40°C
4. The quantity of heat require to change the unit mass of solid to its liquid state, while temp remains constant is called
  - a. Heat
  - b. latent heat
  - c. specific heat
  - d. latent heat of fusion
5. I during constant temp, we feel colder on a day when the relative humidity is
  - a. 85%
  - b. 60%
  - c. 50%
  - d. 25%
6. When a bubble bursts in a boiling liquid, its pressure is
  - a. Equal to the atmospheric pressure
  - b. Larger than atmospheric pressure
  - c. Equal to the vapour pressure in its surroundings
  - d. Larger than the vapour pressure in its surroundings.
7. A artificially heated rooms are dry. Which of the following may be the reason?
  - a. Temp of the air is large
  - b. Pressure of the air is large
  - c. Pressure of the air is low
  - d. Humidity of the air is low.
8. A liquid is boiling in an airtight vessel. Using an exhaust tube, the vapour is slowly removed from the vessel. What happens to the liquid?
  - a. Temp goes down, boiling continues
  - b. Temp goes down, boiling stops
  - c. Temp goes up, boiling continues
  - d. Temp goes up, boiling stops

9. What energy transformation takes place when ice is converted into water?
- Heat energy into kinetic energy
  - kinetic energy into Heat energy
  - Heat energy to latent heat
  - Heat energy to potential energy
10. Unsaturated vapour
- Do not obey gas law
  - Fully obey gas law
  - Obey Boil's law but not Charle's law
  - Partially obey gas law depending on the degree of instauration
11. At boiling point the SVP of a vapour above a liquid is
- Equal to the external pressure
  - Less than external pressure
  - More than external pressure
  - Equal to SVP at room temp
12. Ice is slippery when a man walks on it for
- Its surface is smooth
  - There is no friction
  - Increase of pressure causes melting of ice
  - None of this
13. The boiling point of the solution of a solid in a liquid is
- Greater than that of the solvent
  - Less than that of the solvent
  - Same as that of the solvent .
14. Sublimation refers to
- Vaporization of solid without first becoming a liquid
  - The melting of solid
  - The vaporizing of the liquid
  - The condensation of gas into a liquid
15. When a vapour condenses into liquid
- It absorbs heat
  - It evolves heat
  - Its temperature rises
  - Its temp drops

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