## ST. LAWRENCE HIGH SCHOOL

## A JESUIT CHRISTIAN MINORITY INSTITUTION

- SubjectPhysical science Worksheet- 8Class 7
- Date 8.06.2020
- Chapter:Elements, Compounds And Mixtures
- Answer the following questions (MCQ) :
- Question 1
Q. The one or two letters used to represent elements
answer choices

Yield sign
Coefficient
Symbol
Subscript

- Question 2
Q. Element or compound?
$\mathrm{H}_{2} \mathrm{O}$
answer choices
element
compound
- Question 3
Q. Element or compound?
$\mathrm{O}_{2}$
answer choices

Element
Compound

- Question 4
Q. How many elements are in $\mathrm{C}_{6} \mathrm{H}_{12} \mathrm{O}_{6}$ ?
answer choices

1
2

- Question 5
Q. A physical change is?
answer choices

It changes the object into something different
Objects or substances undergo a change that does not change their chemical composition it is just like a chemical reaction

Paper burning

- Question 6
Q. A chemical change is?
answer choices
a shoe lace cut in half
doesn't change anything
changes the look of an object
change that results in the formation of new chemical substances
- Question 7
Q. The smallest physical unit of an element or compound $\left(\mathrm{O}_{2}\right.$ and $\mathrm{H}_{2} \mathrm{O}$ are both examples of ONE of these)
answer choices

Physical change
Chemical Symbol
Yield Sign
Molecule

- Question 8
Q. The arrow symbol used in a chemical equation
answer choices

Yield sign
Subscript
Coefficient
Chemical formula

- Question 9
Q. A way to show the number and type of elements (atoms) in a molecule or compound with a combination of symbols and numbers.
answer choices

Molecule
Chemical formula
Physical change

## Coefficient

- Question 10
Q. The small number that is written below the symbol and shows how many atoms of each element is present.
answer choices

Compound
Molecule
Coefficient
Subscript

- Question 11
Q. The large number written in front of a chemical formula. It shows the number of molecules.
answer choices
Molecule
Compound
Coeffiecient
Subscript
- Question 12
Q. How many Hydrogen atoms are in $\mathrm{H}_{2} \mathrm{O}$ ?
answer choices
zero
one
two
three
- Question 13
Q. How many atoms of carbon (C) are in $\mathrm{C}_{6} \mathrm{H}_{12} \mathrm{O}_{6}$ ?
- Question 14
Q. Which of the following is a compound?
answer choices

Co
$\mathrm{O}_{2}$
C
$\mathrm{CO}_{2}$

- Question 15
Q. Which of the following is NOT an element?
answer choices
$\mathrm{H}_{2} \mathrm{O}$
$\mathrm{H}_{2}$
$\mathrm{O}_{2}$
Au

