

a) solid

b)gas

ST.LAWRENCE HIGH SCHOOL



JESUIT MINORITY INSTITUTION

CLASS 5- GENERAL SCIENCE- Worksheet 1

OPIC:	STATES OF WATT	ER	DATE: 13.04.2020			
1.	. Which of the following states of matter occur between solid and gas as the energincreased?					
2.	a) Solid In which of the other?	b)liquid following three s	c)gas tates of matter are the particles farthest from each			
	a) Solid	b)liquid	c)gas			
3.	Which of the following is the common name for the solid state of water?					
	a)liquid	b)steam	c)ice			
4.	4. Matter is anything that occupies space and has					
	a)atom	b)weight	c) volume			
5.	. Which state(s) of matter takes the shape of the container they are placed in?					
	a) Liquid and s	olid b) Gas on	ly c) Liquid only			
6.	5. There are three states of matter: solid, liquid and gas. Select one property that liquids and gases have in common.					
	a) They have a	definite shape	b) They cannot be compressed.			
	c) They can flo	W.				
7.	7. Which one of the following statements is correct?					
	a) Gases canno	t be compressed.	b) Solids are easily compressed.			
	c) Gases are easier to compress than solids.					
8.	3. Atoms of different kind forms a molecule of a					
	a) element	b)compound	c)matter			
9.	What state of m	natter will keep th	e same amount but its shape will change?			

c)liquid

10. Molecules are always in a state of						
a)rest	b) motion	c)solid				
11. Intermolecular force in gas is						
a) very weak	b) weak	c)strong				
12. Which one of the following is not a property of a gas?						
a)definite mass	b)no defi	nite shape	c)definite volume			
13. Liquids have the ability to flow because						
a) their particles are arranged in a regular pattern						
b) their particles can slide over each other						
c) there are strong forces of attraction between the particles.						
14. What state of matter has neither a fixed shape nor a fixed volume?						
a) gas	b)solid	c)liquid				
15. Solids have intermolecular space.						
a) more	b) very less	c)moder	ate			

Ms. Beline Jermy Peter