

ST. LAWRENCE HIGH SCHOOL

A Jesuit Christian Minority Institution

Class: X



Worksheet Number: 1

Sub: Physical Science Topic: Inorganic Chemistry

F.M.: 15 Date: 30th April 2020

Answer the following questions:

1. Which one is used to prepare ammonia?

(a) NH₄Cl (b) (NH₄)₂SO₄ (c) NH₄NO₃ (d) Ca(OH)₂

2. An aqueous solution of ammonia behaves as:

(a) acid (b) basic (c) neutral (d) changes as per condition

3. Pure ammonia is collected:

(a) over water (b) over alcohol (c) over ether (d) over mercury

4. NH₃ is a/an

(a) oxidant (b) reductant (c) both (d) none of these

5. Ammonia turns in presence of Nessler's reagent:

(a) black (b) green (c) brown (d) yellow

6. Liquor ammonia contains _____ ammonia.

(a) 25% (b) 30% (C) 35% (d) 45%

7. What is the industrial manufacturing of ammonia known as?

(a) Oswald process (b) Solvey process (c) Bayer process (d) Haber process

8. The reaction of Haber process is a:

(a) replacement reaction (b) synthesis reaction (c) decomposition reaction (d) combination reaction

9. The catalyst used in Haber process is:

(a) MnO₂ (b) platinum dust (c) iron powder (d) copper powder

[1×15=15]

10. Which one is produced in the first step in the manufacturing of urea?

(a) ammonium carbamate (b) ammonium carbonate (c) ammonium sulphate (d) ammonium sulphite

11. Incomplete combustion of H.S produce:

(a) SO_2 (b) H_2SO_4 (c) S (d) both (a) & (C)

12. Which one is used to manufacture the sedative barbiturate?

(a) ammonia (b) HNO_3 (c) H_2S (d) urea

13. H₂S is a poisonous gas because it:

- (a) affects the CNS
- (b) decrease the efficiency of lungs
- (c) causes cerebral haemorrhage
- (d) deactivates the essential cellular proteins.

14. Which reaction is exothermic?

- (a) $N_2 + 3H_2 \rightarrow 2NH_3$
- (b) $N_2 + O_2 \rightarrow 2NO$
- (c) $4NH_3 + 5O_2 \rightarrow 4NO + 6H_2O$
- (d) $2SO_2 + O_2 \rightarrow 2SO_3$

15. When Hydrogen Sulphide is passed through lead nitrate, the colour of the precipitate obtained is:

(a) white (b) brown (c) green (d) black

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