

ST. LAWRENCE HIGH SCHOOL



A JESUIT CHRISTIAN MINORITY INSTITUTION

- Subject Chemistry Answers of Worksheet-7 Class IX
- Date 18.05.2020
- Chapter Acid .bases and salts

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Answer the following questions (MCQ):	(1×15)
Question 1 Q. Metal + Oxygen yields	
Question 2 Q. Metal + Water yields answer choices Metal Oxide + Hydrogen Metal Hydroxide + Hydrogen A + B Metal Oxide Answer A + B	
Question 3 Q. Metal + Steam yields answer choices Metal Oxide + Hydrogen Metal Hydroxide + Hydrogen A + B Metal Oxide Answer A + B	
Question 4 Q. Metal + Acid yields answer choices Salt + Hydrogen Salt + Water + Carbon Dioxide Salt + Water Salt + Alkali Answer Salt + Hydrogen	
Question 5 Q. Acid + Carbonate yields answer choices Salt + Hydrogen	

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Salt + Water + Carbon Dioxide
Salt + Water
Salt + Alkali
Answer Salt + Water + Carbon Dioxide
Question 6
Q.
Acid + Alkali yields
answer choices
Salt + Hydrogen
Salt + Water + Carbon Dioxide
Salt + Water
Salt + Alkali
Answer Salt + Water
Question 7
Q.
Which Salt is form with the following acid- Hydrochloric Acid
answer choices
Chloride
Sulfate
Nitrate
Answer Chloride
Question 8
Q.
Which Salt is form with the following acid- Sulfuric Acid
answer choices
Chloride
Sulphate
Nitrate
Answer Sulphate
Question 9
Q.
Which Salt is form with the following acid- Nitric Acid
answer choices
Chloride
Sulfate
Nitrate
Answer Nitrate
Question 10
What are the products of the equation when you combine Platinum and Water?
answer choices
Platinum
Platinum + Water
No Reaction
Answer No Reaction
Question 11
What is the chemical formulae for Hydrochloric acid?
answer choices
H_2SO_4
HNO<sub>3</sub>
HCI
Answer HCI
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Question 12
 What is the chemical formulae for Nitric acid?
 answer choices
 H_2SO_4
 HNO_3
 HCI
 Answer HNO<sub>3</sub>
  Question 13
 Q.
 What is the chemical formulae for Sulfuric acid?
 answer choices
 H_2SO_4
 HNO_3
 HCI
 Answer H<sub>2</sub>SO<sub>4</sub>
  Question 14
 Q.
 What is the word equation for Neutralization?
 answer choices
 Acid + Alkali ---> Salt + Water
 Acid + Carbonate ---> Salt + Water + Carbon Dioxide
 Acid + Alkali ---> Salt + Water + Carbon Dioxide
 Acid + Carbonate ---> Salt + Water
 Answer Acid + Alkali ---> Salt + Water
  Question 15
 Q.
 What are the two products of Neutralization?
 answer choices
 Salt + Water
 Salt + Water + Carbon Dioxide
Salt + Water + Hydrogen
 AnswerSalt + Water
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