# ST. LAWRENCE HIGH SCHOOL A JESUIT CHRISTIAN MINORITY INSTITUTION 

Sub: Physical Science
Duration: $\mathbf{4 0} \mathbf{~ m i n}$

Class: 8
Worksheet 57
LANGUAGE OF CHEMISTRY

Date: 07.07.20
Full Marks: 15

## Choose the Correct options:

$1-\mathrm{KClO}_{3} \rightarrow \mathrm{KCl}+3 \mathrm{O}_{2}$
(A) Is balanced
(B) Not balanced
(C) Incomplete
(D) None of the above
$2-\mathrm{N}_{2}+\mathrm{H}_{2} \rightarrow 2 \mathrm{NH}_{3}$
(A) Is balanced
(B) Not balanced
(C) Incomplete
(D) None of the above
$3-\mathrm{Na}_{2} \mathrm{CO}_{3}+\mathrm{HCl} \longrightarrow \mathrm{NaCl}+\mathrm{H}_{2} \mathrm{O}$
(A) Is balanced
(B) Not balanced
(C) Incomplete
(D) None of the above
$4-2 \mathrm{Mg}+\mathrm{O}_{2} \rightarrow 2 \mathrm{MgO}$
(A) Is balanced
(B) Not balanced
(C) Incomplete
(D) None of the above
$5-2 \mathrm{C}+\mathrm{O}_{2} \rightarrow 2 \mathrm{CO}_{2}$
(A) Is balanced
(B) Not balanced
(C) Incomplete
(D) None of the above

6-Which of the following(s) can be included in a chemical equation
(A) Physical states
(B) Temperature
(C) Catalyst
(D) All of the above

7-Chemical reactions involve
(A) the breaking of bonds
(B) the making of bonds
(C) the breaking and making of bonds
(D) the shifting of bonds

8-A balanced chemical equation is in accordance with
(A) Avogadro's law
(B) Law of multiple proportion
(C) Law of conservation of mass
(D) Law of gaseous volumes

9- The equation
$\mathrm{xMg}+\mathrm{CO}_{2} \rightarrow \mathrm{yMgO}+\mathrm{C}$
The values of $x$ and $y$ are
(A) 3 and 5
(B) 2 and 6
(C) 2 and 2
(D) 3 and 2

10- In the following equation:
$\mathrm{H}_{2}+\mathrm{Cl}_{2} \rightarrow \mathrm{xHCl}$, the value of x is
(A) 1
(B) 2
(C) 3
(D) 4

11- In the balanced equation $\mathrm{aNa}_{2} \mathrm{CO}_{3}+\mathrm{bHCl} \rightarrow \mathrm{cNaCl}+\mathrm{dH}_{2} \mathrm{O}+\mathrm{eCO}_{2}$
-The value of $a, b, c, d, e$ are respectively -
(A) 1, 1,2,3,2
(B) $1,2,2,1,1$
(C) $1,3,2,3,2$
(D) $1,2,2,3,1$

12-Which of the following reactions is balanced
(A) $\mathrm{NaHCO}_{3} \rightarrow \mathrm{Na}_{2} \mathrm{CO}_{3},+\mathrm{H}_{2} \mathrm{O}+\mathrm{CO}_{2}$
(B) $2 \mathrm{C}_{4} \mathrm{H}_{10}+12 \mathrm{O}_{2} \rightarrow 8 \mathrm{CO}_{2}+10 \mathrm{H}_{2} \mathrm{O}$
(C) $2 \mathrm{Al}+4 \mathrm{H}_{2} \mathrm{O} \rightarrow 2 \mathrm{Al}(\mathrm{OH})_{3}+3 \mathrm{H}_{2}$
(D) $4 \mathrm{NH}_{3}+5 \mathrm{O}_{2} \rightarrow 4 \mathrm{NO}+6 \mathrm{H}_{2} \mathrm{O}$

13-The equation $-\mathrm{xH}_{2}+\mathrm{O}_{2} \rightarrow \mathrm{yH}_{2} \mathrm{O}$ The values of x and y are
(A) 3 and 5
(B) 4 and 6
(C) 2 and 2
(D) 7 and 5

14-Which of the following statements is correct
(A) A chemical equation tells us about the substances involved in a reaction.
(B) A chemical equation informs us about the symbols and formula of the substances involved in a reaction.
(C) A chemical equation tells us about the atoms or molecules of the reactants and products involved in a reaction.
(D) All are correct.

15-In the reaction $\mathrm{xPb}\left(\mathrm{NO}_{3}\right)_{2} \rightarrow \mathrm{yPbO}+\mathrm{zNO}_{2}+\mathrm{O}_{2} \mathrm{x}, \mathrm{y}$ and z are -
(A) $1,1,2$
(B) $2,2,4$
(C) $1,2,4$
(D) $4,2,2$

