

## ST. LAWRENCE HIGH SCHOOL



A JESUIT CHRISTIAN MINORITY INSTITUTION

Sub: Physical Science Duration: 40 min	Class Worksheet		Date: 20.03.20 Full Marks: 15
2. Physical and chemical changes/Energy change and physical chemical			
changes occurring together			
Choose the Correct options:			
1. Melting of was is a			
Ans (a)Physical Change		(c)Chemical change	(d) Irreversible change
2. Burning of the wax is a			
Ans (a)Physical Change	(b)Periodic change	(c)Chemical change	(d) Reversible change
3. Heat is a form of			
e e e e e e e e e e e e e e e e e e e	(b) Temperature	(c) Energy	(d) Matter
4. Heat is given out during all			
change	(b) Endothermic change	(c) Reversible change	(d) Undesirable change
5. Heat is absorbed duri	0		
change	(b) Periodic change	change	(d) Undesirable change
6. Formation of a heterogeneous mixture is an example of			
Ans (a) Exothermic change (b) Chemical change (c) Endothermic change (d) None			
7. Melting is an exampl			(1) No
	(b) Chemical	(c) Endothermic	(d) None of
change	change	change	these
8. Freezing of water is an example of			
Ans (a) Exothermic change (b) Chemical change (c) Endothermic change (d) None 9. What is the energy possessed by molecules in the different states of matter			
		(c) Chemical energy	
	Energy	(c) Chemical energy	energy
10. Glucose added to water is an example of			
Ans (a) Exothermic change (b) Chemical change (c) Endothermic change (d) None 11. Adding concentrated hydrochloric acid to water is a			
6	•		(d) None of
Ans (a) Exothermic change	(b) Periodic change	(c) Endothermic change	(d) None of these
12. Which energy of a s			(d) Electrical
Ans (a) Kinetic Energy	(b) Potential Energy	(c) Bond energy	(d) Electrical energy
13. Photosynthesis is an	-		(1) No
Ans (a) Exothermic change	(b) Periodic change	(c) Endothermic change	(d) None of these
14. Slaking of lime is an example of			
Ans (a) Exothermic change	(b) Periodic change	(c) Endothermic change	(d) None of these
15. Surplus of deficit of energy in reactants and products of chemical change decide whether the change will be			
Ans (a) Exothermic	(b) Periodic change	(c) Endothermic	(d) Both (a) and
change	(,, , , , , , , , , , , , , , , , , , ,	change	(c)