## ST. LAWRENCE HIGH SCHOOL



## A JESUIT CHRISTIAN MINORITY INSTITUTION

## SOLUTION<u>-33(CLASS-11)</u> TOPIC- PERIODIC PROPERTIES SUBTOPIC-BASIC CONCEPT



SUBJECT – CHEMISTRY DURATION – 30 mins

F.M. - 15 DATE -17.08.20

1. Find the set of Bridge elements-
(a) Li, Mg
(b) H, He
(c) B, C
(d) He, Li
Ans. a
2. Which among the following is the most electopositive metal-
(a) Cs
(b) Na
(c) Ti
(d) Fe
Ans. a
3. The correct order for the electronegativity among I, Br and F-
(a) I>Br>F
(b) F >Br > I
(c) Br > I >F
(d) Br >I <f< td=""></f<>
Ans. b
4. Which among the following is not a determining factor of Electronegativity-
(a) Oxidation state
(b) Hybridization state
(c) Screening effect
(d) Atomic radius
Ans c

(	a) H
(	b) F
(	c) I
(	d) C
F	Ans. a
6	6. The reducing property of along the group-
(	a) Decreases
(	b) Increases
(	c) Can't be predicted
(	d) Remains unchanged
F	Ans. b
7	7. The correct order of electron affinity is-
(	a) Cl > F >Br>l
(	b) F > O > Cl > Br
(	c) F > Cl > Br > O
(	d) O > F > Cl > Br
F	Ans. a
8	3. The pair of elements those are wrongly matched-
(	a) Ar & Xe
(	b) I & CI
(	c) Na & K
(	d) Cs & At
F	Ans. d
9	). Which one is the most acidic among these?
(	a) MgO
(	b) Cl2O7
(	c) Al <sub>2</sub> O <sub>3</sub>
(	d) Na <sub>2</sub> O
F	Ans. b
1	0. Which one will have the highest Secondionisation energy?
1	a) 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>1</sup>

5. Which of the following has the least electronegativity-

(b)  $1s^2 2s^2 2p^4$ (c)  $1s^2 2s^2 2p^6$ (d)  $1s^2 2s^2 2p^6 3s1$ Ans. d 11. Compounds having incomplete octet-(a) NH3 (b) BF3 (c) H2O (d) CH4 Ans. b 12. Which period contains maximum number of elements? (a) 5 (b) 7 (c) 1 (d) 6 Ans. b 13. Which of the following is not a periodic property? (a)Metallic character (b) Electronegativity (c) Oxidizing power (d)Hardness Ans. d 14. Which among the following is a chalcogens? (a) O, Te, Se (b) Cu, K, Na (c) Au, Pt, W (d) Li, Na, B Ans. a 15. Which among the following has the maximum valency? (a) Os (b) Th (c) Fr

(d) At

Ans. a

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