



ST. LAWRENCE HIGH SCHOOL

A JESUIT CHRISTIAN MINORITY INSTITUTION

WORKSHEET-47(CLASS-12)

TOPIC- ALDEHYDE AND KETONE

SUBTOPIC- PREPARATION AND CHEMICAL REACTIONS

SUBJECT – CHEMISTRY

DURATION – 30 mins



F.M. - 15

DATE -01.08.20

1.1 During the nucleophilic addition reaction of an aldehyde or Ketone the change in hybridization state of the carbonyl atom changes-

a) sp^2 to sp^3 b) sp^3 to sp^2 c) sp^2 to sp d) sp to sp^3

1.2 Wolff Kishner reduction of a ketone is carried out in the presence of which of the following?

a) H_2 and Pt as catalyst b) Glycol with KOH c) Zn-Hg with HCl d) $LiAlH_4$

1.3 The factor/factors that affect the rate of a chemical reaction of an aldehyde or a ketone is / are-

a) Electronic effect b) steric effect c) both a and b d) none of these

1.4 The hydration of an aldehyde or a ketone is reversible due to-

a) Slowest step b) Formation of geminal diol c) Formation of vicinal diol d) Tautomerism

1.5 Ammonia reacts with aldehyde or ketone to form-

a) Imine b) Urea c) Amide d) Amine

1.6 Reaction of ethyl ethanoate with limited supply of CH_3MgI followed by hydrolysis gives-

a) Ethanol b) n-propyl alcohol c) Acetaldehyde d) Isopropyl alcohol

1.7 Both aldehyde and ketone can be identified by-

a) Tollens reagent b) Sodium bisulphite addition c) Brady's reagent d) All of these

1.8 Aldehyde and ketone are reduced to form-

a) Alcohol b) Ether c) Alkane d) alkyne

1.9 A strong base can abstract an α -hydrogen from –

a) Amine b) Both c and d c) Ketone d) Ether

1.10 Reduction of aldehydes and ketones into hydrocarbons using Hydrazine, KOH and under heating condition-

a) Cope reduction b) Dow reduction c) Wolff-Kishner reduction d) Clemmensen reduction

1.11 Chloral forms stable geminal-di-ol, due to-

a) Inter molecular H-bond formation b) Intra molecular H-bond formation c) Inductive effect
d) Hyperconjugation

1.12 Aldehydes are chemically more reactive than the ketones due to-

a) Electrophilic character at the carbonyl carbon b) Steric factor c) both a and b d) none of these

1.13 Semicarbazide on treatment with Aldehyde forms-

a) Salicylaldehyde b) Semicarbazene c) Semicarbazone d) Semicarbazine

1.14 Aldehyde and ketone both undergo the type of chemical reaction/reactions-

a) Condensation b) rearrangement c) addition d) all of these

1.15 The geometry of the carbonyl group is-

a) Trigonal planar b) Pyramidal c) Square planar d) Linear

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